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THE SYNTHESIS OF 2,5-DIHYDROXYBENZO-1,3,2-DITHIAZOLYLIUM AND 2,7-DICARBONYL NAPHTHA-1,3,2-DITHIAZOLYLIUM CATIONS

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Keywords: Cycloaddition; dithionitronium cation

The cycloaddition chemistry of the dithionitronium cation (SNS⁺) has led to the synthesis of dithiadiazolyl (CNSNS, CNSSN) and dithiazolyl (CSNSC) ring systems from the corresponding nitriles and alkynes respectively. Olefins are isolobal with both alkynes and nitriles and can undergo cycloaddition with one or two stoichiometric aliquots of olefin because of the two mutually perpendicular π -manifolds of SNS⁺. The chemistry of SNS⁺ with extended π -systems has not yet been investigated beyond the case of benzene, where an electrophilic substitution results in the sulfur protonated substitution product, likely $C_6H_5SNSH^{+,4}$ Wolmershauer first described the synthesis of benzofused 1,3,2-dithiazolylium chloride salt on a preparative scale, which was characterized by x-ray crystallography.

Preliminary investigations of the extended π -systems of 1,4-benzoquinone and 1,4-naphthoquinone with SNS⁺ have been shown by x-ray crystallography to be 2,5-dihydroxybenzo-1,3,2-dithiazolylium (1⁺ \mathbf{AsF}_6^-) and 2,7-dicarbonylnaphtha-1,3,2-dithiazolylium (2⁺ \mathbf{AsF}_6^-) salts (Figure 1). The salt (1) has been characterized by multinuclear NMR {¹H, ¹³C, ¹⁴N, ¹⁹F, HMQC, COSY}, infrared spectroscopy, and supported by DFT calculations (B3LYP/6-311++G**). The intermediate (3) was found to be thermodynamically unfavorable in the gas phase and suggests that it will not be observed in solution or solid state. Both salts (1) and (2) show a similar layered motif, forming columns in the solid state, and few contacts between the anion and the $C_2S_2N^+$ ring are observed in the case of (1).

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FIGURE 1 Dithiazolylium salts.

The simplicity of these reactions allows easy access to new derivatives of benzo-fused 1,3,2-dithiazolylium cations and subsequent reduction may yield potentially new, useful synthetic materials with interesting magnetic properties.

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